

# Succinic acid, 5-bromo-2-methoxybenzyl pentyl ester

<b>Inchi:</b>	InChI=1S/C17H23BrO5/c1-3-4-5-10-22-16(19)8-9-17(20)23-12-13-11-14(18)6-7-15(13)2
<b>InchiKey:</b>	FLNASCFRQJYELS-UHFFFAOYSA-N
<b>Formula:</b>	C17H23BrO5
<b>SMILES:</b>	CCCCCOC(=O)CCC(=O)OCc1cc(Br)ccc1OC
<b>Mol. weight [g/mol]:</b>	387.27

## Physical Properties

Property code	Value	Unit	Source
gf	-373.11	kJ/mol	Joback Method
hf	-776.11	kJ/mol	Joback Method
hfus	45.10	kJ/mol	Joback Method
hvap	84.19	kJ/mol	Joback Method
log10ws	-5.13		Crippen Method
logp	4.014		Crippen Method
mcvol	264.880	ml/mol	McGowan Method
pc	1728.90	kPa	Joback Method
rinpol	2547.00		NIST Webbook
rinpol	2547.00		NIST Webbook
tb	866.16	K	Joback Method
tc	1077.81	K	Joback Method
tf	559.16	K	Joback Method
vc	1.008	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	770.17	J/molxK	866.16	Joback Method
cpg	783.38	J/molxK	901.44	Joback Method
cpg	795.47	J/molxK	936.71	Joback Method
cpg	806.45	J/molxK	971.99	Joback Method
cpg	816.32	J/molxK	1007.26	Joback Method
cpg	825.08	J/molxK	1042.54	Joback Method
cpg	832.75	J/molxK	1077.81	Joback Method
dvisc	0.0003352	Paxs	559.16	Joback Method

dvisc	0.0002143	Paxs	610.33	Joback Method
dvisc	0.0001468	Paxs	661.49	Joback Method
dvisc	0.0001062	Paxs	712.66	Joback Method
dvisc	0.0000802	Paxs	763.83	Joback Method
dvisc	0.0000627	Paxs	814.99	Joback Method
dvisc	0.0000505	Paxs	866.16	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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