

# Acetic acid, trifluoro, 1-methylheptyl ester

**Inchi:** InChI=1S/C10H17F3O2/c1-3-4-5-6-7-8(2)15-9(14)10(11,12)13/h8H,3-7H2,1-2H3  
**InchiKey:** OSUSEXASYGZJOJ-UHFFFAOYSA-N  
**Formula:** C10H17F3O2  
**SMILES:** CCCCCC(C)OC(=O)C(F)(F)F  
**Mol. weight [g/mol]:** 226.24  
**CAS:** 332-83-2

## Physical Properties

Property code	Value	Unit	Source
gf	-784.63	kJ/mol	Joback Method
hf	-1096.89	kJ/mol	Joback Method
hfus	22.75	kJ/mol	Joback Method
hvap	42.88	kJ/mol	Joback Method
log10ws	-3.65		Crippen Method
logp	3.451		Crippen Method
mcvol	164.510	ml/mol	McGowan Method
pc	1992.98	kPa	Joback Method
rinpol	967.00		NIST Webbook
rinpol	991.50		NIST Webbook
rinpol	991.50		NIST Webbook
rinpol	967.00		NIST Webbook
ripol	1075.00		NIST Webbook
ripol	1075.00		NIST Webbook
tb	498.63	K	Joback Method
tc	660.43	K	Joback Method
tf	263.81	K	Joback Method
vc	0.656	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.78	J/molxK	498.63	Joback Method
cpg	409.38	J/molxK	525.60	Joback Method
cpg	422.39	J/molxK	552.56	Joback Method

cpg	434.81	J/mol×K	579.53	Joback Method
cpg	446.68	J/mol×K	606.50	Joback Method
cpg	458.01	J/mol×K	633.47	Joback Method
cpg	468.80	J/mol×K	660.43	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C332832&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C332832&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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