

3-Methoxy-2,4,5-trifluorobenzoic acid, tetradecyl ester

Inchi:	InChI=1S/C22H33F3O3/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-28-22(26)17-16-18(23)20(2
InchiKey:	ANKSIQFSTWOYQI-UHFFFAOYSA-N
Formula:	C22H33F3O3
SMILES:	CCCCCCCCCCCCCOC(=O)c1cc(F)c(F)c(OC)c1F
Mol. weight [g/mol]:	402.49

Physical Properties

Property code	Value	Unit	Source
gf	-715.10	kJ/mol	Joback Method
hf	-1272.11	kJ/mol	Joback Method
hfus	58.44	kJ/mol	Joback Method
hvap	78.61	kJ/mol	Joback Method
log10ws	-8.27		Crippen Method
logp	6.970		Crippen Method
mvol	315.700	ml/mol	McGowan Method
pc	998.28	kPa	Joback Method
rinpol	2558.00		NIST Webbook
rinpol	2558.00		NIST Webbook
tb	845.88	K	Joback Method
tc	1036.33	K	Joback Method
tf	510.36	K	Joback Method
vc	1.256	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	997.69	J/molxK	845.88	Joback Method
cpg	1014.86	J/molxK	877.62	Joback Method
cpg	1030.93	J/molxK	909.36	Joback Method
cpg	1045.91	J/molxK	941.10	Joback Method
cpg	1059.82	J/molxK	972.84	Joback Method
cpg	1072.66	J/molxK	1004.58	Joback Method
cpg	1084.46	J/molxK	1036.33	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U338771&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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