

# Fumaric acid, 2,4,6-trichlorophenyl 1,1,1-trifluoroprop-2-yl ester

<b>Inchi:</b>	InChI=1S/C13H8Cl3F3O4/c1-6(13(17,18)19)22-10(20)2-3-11(21)23-12-8(15)4-7(14)5-9(
<b>InchiKey:</b>	QICJDILZYKHBSN-NSCUHMNNSA-N
<b>Formula:</b>	C13H8Cl3F3O4
<b>SMILES:</b>	CC(OC(=O)C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	391.55

## Physical Properties

Property code	Value	Unit	Source
gf	-865.34	kJ/mol	Joback Method
hf	-1131.49	kJ/mol	Joback Method
hfus	38.97	kJ/mol	Joback Method
hvap	76.08	kJ/mol	Joback Method
log10ws	-5.42		Crippen Method
logp	4.602		Crippen Method
mcvol	222.880	ml/mol	McGowan Method
pc	1966.56	kPa	Joback Method
rinpol	1920.00		NIST Webbook
rinpol	1920.00		NIST Webbook
tb	801.63	K	Joback Method
tc	1018.08	K	Joback Method
tf	518.44	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	555.29	J/mol×K	801.63	Joback Method
cpg	564.13	J/mol×K	837.70	Joback Method
cpg	572.20	J/mol×K	873.78	Joback Method
cpg	579.53	J/mol×K	909.85	Joback Method
cpg	586.16	J/mol×K	945.93	Joback Method
cpg	592.12	J/mol×K	982.00	Joback Method
cpg	597.44	J/mol×K	1018.08	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405945&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405945&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvpap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinppl:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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