

# Methanone, (4-chlorophenyl)(4-hydroxyphenyl)-

Other names:	4-Hydroxy-4'-chlorobenzophenone 4-Chloro-4'-hydroxybenzophenone
Inchi:	InChI=1S/C13H9ClO2/c14-11-5-1-9(2-6-11)13(16)10-3-7-12(15)8-4-10/h1-8,15H
InchiKey:	RUETVLNXAGWCDS-UHFFFAOYSA-N
Formula:	C13H9ClO2
SMILES:	O=C(c1ccc(O)cc1)c1ccc(Cl)cc1
Mol. weight [g/mol]:	232.66
CAS:	42019-78-3

## Physical Properties

Property code	Value	Unit	Source
gf	-21.70	kJ/mol	Joback Method
hf	-155.69	kJ/mol	Joback Method
hfus	28.70	kJ/mol	Joback Method
hvap	73.89	kJ/mol	Joback Method
log10ws	-3.66		Crippen Method
logp	3.277		Crippen Method
mvol	166.190	ml/mol	McGowan Method
pc	3699.95	kPa	Joback Method
tb	727.10	K	Joback Method
tc	990.16	K	Joback Method
tf	493.20	K	Joback Method
vc	0.569	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	409.69	J/mol×K	727.10	Joback Method
cpg	421.13	J/mol×K	770.94	Joback Method
cpg	431.67	J/mol×K	814.79	Joback Method
cpg	441.45	J/mol×K	858.63	Joback Method
cpg	450.62	J/mol×K	902.48	Joback Method
cpg	459.34	J/mol×K	946.32	Joback Method
cpg	467.77	J/mol×K	990.16	Joback Method

dvisc	0.0002993	Paxs	493.20	Joback Method
dvisc	0.0001539	Paxs	532.18	Joback Method
dvisc	0.0000866	Paxs	571.17	Joback Method
dvisc	0.0000525	Paxs	610.15	Joback Method
dvisc	0.0000338	Paxs	649.13	Joback Method
dvisc	0.0000228	Paxs	688.12	Joback Method
dvisc	0.0000161	Paxs	727.10	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	530.20	K	1.70	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42019783&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42019783&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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