

# N-Cyano-S,S'-dimethyldithioimido carbonate

<b>Other names:</b>	Dimethyl cyanodithioiminocarbonate Dimethyl N-cyanodithioiminocarbonate S,S'-Dimethyl N-cyanodithioiminocarbonate Carbonimidodithioic acid, cyano-, dimethyl ester (Cyanoimino)bis(methylthio)methane Cyanodithioimidocarbonic acid, dimethyl ester Dimethyl cyanimidodithiocarbonate Dimethyl cyanocarbonimidodithioate Dimethyl cyanoiminodithiocarbonate Dimethyl N-cyanoimidodithiocarbonate Imidocarbonic acid, cyanodithio-, dimethyl ester S,S-Dimethyl (N-cyanoimido)dithiocarbonate S,S-Dimethyl cyaniminodithiocarbamate Carbonimidodithioic acid, N-cyano-, dimethyl ester NSC 145987 S,S-Dimethyl cyanoimidodithiocarbonate
<b>Inchi:</b>	InChI=1S/C4H6N2S2/c1-7-4(8-2)6-3-5/h1-2H3
<b>InchiKey:</b>	IULFXBLVJIPESI-UHFFFAOYSA-N
<b>Formula:</b>	C4H6N2S2
<b>SMILES:</b>	CSC(=NC#N)SC
<b>Mol. weight [g/mol]:</b>	146.23
<b>CAS:</b>	10191-60-3

## Physical Properties

Property code	Value	Unit	Source
hf	195.16	kJ/mol	Joback Method
hvap	52.00	kJ/mol	Joback Method
ie	9.10	eV	NIST Webbook
ie	9.40 ± 0.05	eV	NIST Webbook
log10ws	-1.78		Crippen Method
logp	1.549		Crippen Method
mcvol	106.980	ml/mol	McGowan Method
pc	3488.88	kPa	Joback Method
tb	607.12	K	Joback Method
tc	868.96	K	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10191603&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10191603&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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