

# Fumaric acid, isoheptyl 2,3,4,6-tetrachlorophenyl ester

**Inchi:** InChI=1S/C16H16Cl4O4/c1-9(2)4-3-7-23-12(21)5-6-13(22)24-16-11(18)8-10(17)14(19)15  
**InchiKey:** GIRQPJZMYSGNMT-AATRIKPKSA-N  
**Formula:** C16H16Cl4O4  
**SMILES:** CC(C)CCCOC(=O)C=CC(=O)Oc1c(Cl)cc(Cl)c(Cl)c1Cl  
**Mol. weight [g/mol]:** 414.11

## Physical Properties

Property code	Value	Unit	Source
gf	-280.05	kJ/mol	Joback Method
hf	-623.54	kJ/mol	Joback Method
hfus	48.72	kJ/mol	Joback Method
hvap	91.56	kJ/mol	Joback Method
log10ws	-6.35		Crippen Method
logp	5.741		Crippen Method
mcvol	272.080	ml/mol	McGowan Method
pc	1627.22	kPa	Joback Method
rinpol	2669.00		NIST Webbook
rinpol	2669.00		NIST Webbook
tb	918.10	K	Joback Method
tc	1145.61	K	Joback Method
tf	590.50	K	Joback Method
vc	1.042	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	711.78	J/molxK	918.10	Joback Method
cpg	752.12	J/molxK	1107.69	Joback Method
cpg	745.96	J/molxK	1069.77	Joback Method
cpg	738.87	J/molxK	1031.86	Joback Method
cpg	730.82	J/molxK	993.94	Joback Method
cpg	721.80	J/molxK	956.02	Joback Method
cpg	757.36	J/molxK	1145.61	Joback Method
dvisc	0.0000444	Paxs	918.10	Joback Method

dvisc	0.0000549	Paxs	863.50	Joback Method
dvisc	0.0000699	Paxs	808.90	Joback Method
dvisc	0.0000921	Paxs	754.30	Joback Method
dvisc	0.0001266	Paxs	699.70	Joback Method
dvisc	0.0001838	Paxs	645.10	Joback Method
dvisc	0.0002859	Paxs	590.50	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348197&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348197&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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