

# 4-(4-Chloro-2-methylphenoxy)butyric acid, isobutyl ester

Inchi:	InChI=1S/C15H21ClO3/c1-11(2)10-19-15(17)5-4-8-18-14-7-6-13(16)9-12(14)3/h6-7,9,11
InchiKey:	YWYRKYNYHRGXSU-UHFFFAOYSA-N
Formula:	C15H21ClO3
SMILES:	Cc1cc(Cl)ccc1OCCCC(=O)OCC(C)C
Mol. weight [g/mol]:	284.78

## Physical Properties

Property code	Value	Unit	Source
gf	-184.72	kJ/mol	Joback Method
hf	-537.38	kJ/mol	Joback Method
hfus	32.52	kJ/mol	Joback Method
hvap	68.15	kJ/mol	Joback Method
log10ws	-4.30		Crippen Method
logp	4.007		Crippen Method
mcvol	224.000	ml/mol	McGowan Method
pc	1800.03	kPa	Joback Method
rinpol	2379.00		NIST Webbook
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tb	714.94	K	Joback Method
tc	919.70	K	Joback Method
tf	419.58	K	Joback Method
vc	0.853	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.91	J/molxK	714.94	Joback Method
cpg	618.29	J/molxK	749.07	Joback Method
cpg	632.75	J/molxK	783.19	Joback Method
cpg	646.28	J/molxK	817.32	Joback Method
cpg	658.90	J/molxK	851.45	Joback Method
cpg	670.62	J/molxK	885.57	Joback Method
cpg	681.44	J/molxK	919.70	Joback Method
dvisc	0.0008995	Paxs	419.58	Joback Method

dvisc	0.0004991	Paxs	468.81	Joback Method
dvisc	0.0003098	Paxs	518.03	Joback Method
dvisc	0.0002089	Paxs	567.26	Joback Method
dvisc	0.0001500	Paxs	616.49	Joback Method
dvisc	0.0001131	Paxs	665.71	Joback Method
dvisc	0.0000887	Paxs	714.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415078&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415078&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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