

Glutaric acid, 2,2,3,3,4,4,5,5-octafluoropentyl hex-4-yn-3-yl ester

Inchi: InChI=1S/C16H18F8O4/c1-3-6-10(4-2)28-12(26)8-5-7-11(25)27-9-14(19,20)16(23,24)15
InchiKey: NNVKMOXZPMEGOK-UHFFFAOYSA-N
Formula: C16H18F8O4
SMILES: CC#CC(CC)OC(=O)CCCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
Mol. weight [g/mol]: 426.30

Physical Properties

Property code	Value	Unit	Source
gf	-1736.04	kJ/mol	Joback Method
hf	-2196.56	kJ/mol	Joback Method
hfus	41.24	kJ/mol	Joback Method
hvap	60.47	kJ/mol	Joback Method
log10ws	-5.41		Crippen Method
logp	4.216		Crippen Method
mvol	256.740	ml/mol	McGowan Method
pc	1285.59	kPa	Joback Method
rinpol	1684.00		NIST Webbook
rinpol	1684.00		NIST Webbook
tb	710.65	K	Joback Method
tc	881.82	K	Joback Method
tf	502.48	K	Joback Method
vc	1.040	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	758.36	J/mol×K	710.65	Joback Method
cpg	771.71	J/mol×K	739.18	Joback Method
cpg	784.24	J/mol×K	767.71	Joback Method
cpg	795.99	J/mol×K	796.24	Joback Method
cpg	806.99	J/mol×K	824.77	Joback Method
cpg	817.30	J/mol×K	853.30	Joback Method
cpg	826.94	J/mol×K	881.82	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U393971&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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