

Carbanolate

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|-----------------------------|---|
| Other names: | Phenol, 2-chloro-4,5-dimethyl-, methylcarbamate Carbamic acid, methyl-, 6-chloro-3,4-xylyl ester Banol 3,4-Dimethyl-6-chlorophenyl N-methylcarbamate 6-Chloro-3,4-dimethylphenyl N-methylcarbamate 6-Chloro-3,4-xylyl methylcarbamate Banol tuco sok Carbamic acid, methyl-, (2-chloro-4,5-dimethyl)phenyl ester Carbamic acid, methyl-, 2-chloro-4,5-xylyl ester 2-Chloro-4,5-dimethylphenyl methylcarbamate Chloroxylam 6-Chloro-3,4-xylene N-methylcarbamate 2-Chloro-4,5-xylyl N-methylcarbamate 6-Chloro-3,4-xylyl N-methylcarbamate OMS-174 SOK U 12927 U-17004 UPJOHN U-12,927 3,4-Xylyl-6-chloro-N-methylcarbamate |
| Inchi: | InChI=1S/C10H12ClNO2/c1-6-4-8(11)9(5-7(6)2)14-10(13)12-3/h4-5H,1-3H3,(H,12,13) |
| InchiKey: | QRTXZGIQTYDABO-UHFFFAOYSA-N |
| Formula: | C10H12ClNO2 |
| SMILES: | CNC(=O)Oc1cc(C)c(C)cc1Cl |
| Mol. weight [g/mol]: | 213.66 |
| CAS: | 671-04-5 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|--------|----------------|
| gf | -39.62 | kJ/mol | Joback Method |
| hf | -254.68 | kJ/mol | Joback Method |
| hfus | 26.61 | kJ/mol | Joback Method |
| hvap | 62.09 | kJ/mol | Joback Method |
| log10ws | -3.59 | | Crippen Method |
| logp | 2.675 | | Crippen Method |
| mcvol | 157.660 | ml/mol | McGowan Method |
| pc | 2859.68 | kPa | Joback Method |

| | | | |
|----|--------|----------------------|---------------|
| tb | 633.71 | K | Joback Method |
| tc | 852.99 | K | Joback Method |
| tf | 421.18 | K | Joback Method |
| vc | 0.596 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 370.82 | J/mol×K | 633.71 | Joback Method |
| cpg | 382.84 | J/mol×K | 670.26 | Joback Method |
| cpg | 394.16 | J/mol×K | 706.80 | Joback Method |
| cpg | 404.79 | J/mol×K | 743.35 | Joback Method |
| cpg | 414.73 | J/mol×K | 779.89 | Joback Method |
| cpg | 423.99 | J/mol×K | 816.44 | Joback Method |
| cpg | 432.56 | J/mol×K | 852.99 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C671045&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |

tb: Normal Boiling Point Temperature
tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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