

# Phthalic acid, decyl 3,4,5-trichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C24H27Cl3O4/c1-2-3-4-5-6-7-8-11-14-30-23(28)18-12-9-10-13-19(18)24(29)3
<b>InchiKey:</b>	UHCQTCFFSZGMOQ-UHFFFAOYSA-N
<b>Formula:</b>	C24H27Cl3O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)c1ccccc1C(=O)Oc1cc(Cl)c(Cl)c(Cl)c1
<b>Mol. weight [g/mol]:</b>	485.83

## Physical Properties

Property code	Value	Unit	Source
gf	-166.13	kJ/mol	Joback Method
hf	-648.33	kJ/mol	Joback Method
hfus	62.61	kJ/mol	Joback Method
hvap	107.68	kJ/mol	Joback Method
log10ws	-9.62		Crippen Method
logp	8.163		Crippen Method
mcvol	353.100	ml/mol	McGowan Method
pc	1157.71	kPa	Joback Method
rinpol	3387.00		NIST Webbook
rinpol	3387.00		NIST Webbook
tb	1086.67	K	Joback Method
tc	1331.19	K	Joback Method
tf	697.24	K	Joback Method
vc	1.359	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1074.42	J/molxK	1086.67	Joback Method
cpg	1084.30	J/molxK	1127.42	Joback Method
cpg	1092.67	J/molxK	1168.18	Joback Method
cpg	1099.58	J/molxK	1208.93	Joback Method
cpg	1105.09	J/molxK	1249.68	Joback Method
cpg	1109.23	J/molxK	1290.44	Joback Method
cpg	1112.07	J/molxK	1331.19	Joback Method
dvisc	0.0001366	Paxs	697.24	Joback Method

dvisc	0.0000871	Paxs	762.14	Joback Method
dvisc	0.0000596	Paxs	827.05	Joback Method
dvisc	0.0000431	Paxs	891.96	Joback Method
dvisc	0.0000326	Paxs	956.86	Joback Method
dvisc	0.0000255	Paxs	1021.77	Joback Method
dvisc	0.0000206	Paxs	1086.67	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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