

# 5-Ethenyl-5-(1-methyl-3-butenyl)-hexahydropyrim

<b>Other names:</b>	2,4,6(1H,3H,5H)-Pyrimidinetrione, 5-ethenyl-5-(1-methyl-3-butenyl)-Vinylbital M (OH, -H <sub>2</sub> O)
<b>Inchi:</b>	InChI=1S/C11H14N2O3/c1-4-6-7(3)11(5-2)8(14)12-10(16)13-9(11)15/h4-5,7H,1-2,6H2,3
<b>InchiKey:</b>	DMEKNBFGYSDJIZ-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>11</sub> H <sub>14</sub> N <sub>2</sub> O <sub>3</sub>
<b>SMILES:</b>	C=CCC(C)C1(C=C)C(=O)NC(=O)NC1=O
<b>Mol. weight [g/mol]:</b>	222.24
<b>CAS:</b>	131147-55-2

## Physical Properties

Property code	Value	Unit	Source
gf	41.59	kJ/mol	Joback Method
hf	-292.71	kJ/mol	Joback Method
hfus	21.41	kJ/mol	Joback Method
hvap	63.89	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	0.737		Crippen Method
mcvol	171.060	ml/mol	McGowan Method
pc	3166.83	kPa	Joback Method
rinp	1995.00		NIST Webbook
rinp	1995.00		NIST Webbook
tb	764.35	K	Joback Method
tc	1026.96	K	Joback Method
tf	641.21	K	Joback Method
vc	0.633	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	499.42	J/mol×K	764.35	Joback Method
cpg	516.86	J/mol×K	808.12	Joback Method
cpg	533.36	J/mol×K	851.89	Joback Method
cpg	548.95	J/mol×K	895.66	Joback Method
cpg	563.65	J/mol×K	939.42	Joback Method

cpg	577.49	J/mol×K	983.19	Joback Method
cpg	590.50	J/mol×K	1026.96	Joback Method

## Sources

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C131147552&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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