

# Fumaric acid, 2-methoxyphenyl 2,2,3,3,4,4,5,5-octafluoropentyl ester

**Inchi:** InChI=1S/C16H12F8O5/c1-27-9-4-2-3-5-10(9)29-12(26)7-6-11(25)28-8-14(19,20)16(23,24)17  
**InchiKey:** JUJLJYRDJYTD-VOTSOKGWSA-N  
**Formula:** C16H12F8O5  
**SMILES:** COc1ccccc1OC(=O)C=CC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F  
**Mol. weight [g/mol]:** 436.25

## Physical Properties

Property code	Value	Unit	Source
gf	-1858.40	kJ/mol	Joback Method
hf	-2253.52	kJ/mol	Joback Method
hfus	36.69	kJ/mol	Joback Method
hvap	64.02	kJ/mol	Joback Method
log10ws	-4.81		Crippen Method
logp	3.871		Crippen Method
mvol	243.150	ml/mol	McGowan Method
pc	1472.49	kPa	Joback Method
rinpol	1923.00		NIST Webbook
rinpol	1923.00		NIST Webbook
tb	760.33	K	Joback Method
tc	945.01	K	Joback Method
tf	467.47	K	Joback Method
vc	0.975	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	719.73	J/molxK	760.33	Joback Method
cpg	731.00	J/molxK	791.11	Joback Method
cpg	741.43	J/molxK	821.89	Joback Method
cpg	751.05	J/molxK	852.67	Joback Method
cpg	759.92	J/molxK	883.45	Joback Method
cpg	768.10	J/molxK	914.23	Joback Method
cpg	775.64	J/molxK	945.01	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405927&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405927&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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