

# Propionamide, 2,3-dichloro-N-3-methylbutyl-

<b>Inchi:</b>	InChI=1S/C8H15Cl2NO/c1-6(2)3-4-11-8(12)7(10)5-9/h6-7H,3-5H2,1-2H3,(H,11,12)
<b>InchiKey:</b>	KFBUGZALPNZPDM-UHFFFAOYSA-N
<b>Formula:</b>	C8H15Cl2NO
<b>SMILES:</b>	CC(C)CCNC(=O)C(Cl)CCl
<b>Mol. weight [g/mol]:</b>	212.12

## Physical Properties

Property code	Value	Unit	Source
gf	-51.79	kJ/mol	Joback Method
hf	-309.60	kJ/mol	Joback Method
hfus	24.52	kJ/mol	Joback Method
hvap	54.58	kJ/mol	Joback Method
log10ws	-2.31		Crippen Method
logp	1.995		Crippen Method
mcvol	159.610	ml/mol	McGowan Method
pc	2581.96	kPa	Joback Method
rinpol	1630.00		NIST Webbook
rinpol	1630.00		NIST Webbook
tb	560.46	K	Joback Method
tc	756.90	K	Joback Method
tf	312.35	K	Joback Method
vc	0.611	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	358.13	J/molxK	560.46	Joback Method
cpg	370.44	J/molxK	593.20	Joback Method
cpg	382.10	J/molxK	625.94	Joback Method
cpg	393.13	J/molxK	658.68	Joback Method
cpg	403.54	J/molxK	691.42	Joback Method
cpg	413.36	J/molxK	724.16	Joback Method
cpg	422.61	J/molxK	756.90	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U415225&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U415225&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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