

L-Valine, N-pentafluorobenzoyl-, heptadecyl ester

Inchi:	InChI=1S/C29H44F5NO3/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-38-29(37)27(2
InchiKey:	HXVCZIYSLVGVQX-UHFFFAOYSA-N
Formula:	C29H44F5NO3
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(NC(=O)c1c(F)c(F)c(F)c(F)c1F)C(C)C
Mol. weight [g/mol]:	549.66

Physical Properties

Property code	Value	Unit	Source
gf	-994.82	kJ/mol	Joback Method
hf	-1757.73	kJ/mol	Joback Method
hfus	80.80	kJ/mol	Joback Method
hvap	103.21	kJ/mol	Joback Method
log10ws	-11.00		Crippen Method
logp	8.551		Crippen Method
mvol	423.550	ml/mol	McGowan Method
pc	680.65	kPa	Joback Method
rinpol	3208.00		NIST Webbook
rinpol	3208.00		NIST Webbook
tb	1090.30	K	Joback Method
tc	1377.25	K	Joback Method
tf	653.31	K	Joback Method
vc	1.694	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1494.57	J/molxK	1090.30	Joback Method
cpg	1513.40	J/molxK	1138.12	Joback Method
cpg	1529.68	J/molxK	1185.95	Joback Method
cpg	1543.55	J/molxK	1233.77	Joback Method
cpg	1555.13	J/molxK	1281.60	Joback Method
cpg	1564.55	J/molxK	1329.42	Joback Method
cpg	1571.95	J/molxK	1377.25	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U346616&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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