

# Uracil, 1-methyl-

<b>Other names:</b>	1-Methyluracil 2,4(1H,3H)-Pyrimidinedione, 1-methyl-
<b>Inchi:</b>	InChI=1S/C5H6N2O2/c1-7-3-2-4(8)6-5(7)9/h2-3H,1H3,(H,6,8,9)
<b>InchiKey:</b>	XBCXJKGHPABGSD-UHFFFAOYSA-N
<b>Formula:</b>	C5H6N2O2
<b>SMILES:</b>	Cn1ccc(=O)[nH]c1=O
<b>Mol. weight [g/mol]:</b>	126.11
<b>CAS:</b>	615-77-0

## Physical Properties

Property code	Value	Unit	Source
ie	9.00 ± 0.10	eV	NIST Webbook
log10ws	-0.80		Aqueous Solubility Prediction Method
log10ws	-0.81		Estimated Solubility Method
logp	-1.408		Crippen Method
mvol	89.250	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	121.70 ± 4.00	kJ/mol	385.50	NIST Webbook
hsubt	112.50 ± 2.60	kJ/mol	398.00	NIST Webbook
hsubt	105.00 ± 8.00	kJ/mol	457.50	NIST Webbook

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C615770&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

## Legend

<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

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