

# 3-Hydroxytyramine, N,O,O'-tris(trifluoroacetyl)-

<b>Other names:</b>	Dopamine, TFA
<b>Inchi:</b>	InChI=1S/C14H8F9NO5/c15-12(16,17)9(25)24-4-3-6-1-2-7(28-10(26)13(18,19)20)8(5-6)
<b>InchiKey:</b>	ROWYZKLUWDEHTH-UHFFFAOYSA-N
<b>Formula:</b>	C14H8F9NO5
<b>SMILES:</b>	O=C(NCCc1ccc(OC(=O)C(F)(F)F)c(OC(=O)C(F)(F)F)c1)C(F)(F)F
<b>Mol. weight [g/mol]:</b>	441.20

## Physical Properties

Property code	Value	Unit	Source
gf	-2091.99	kJ/mol	Joback Method
hf	-2458.65	kJ/mol	Joback Method
hfus	43.03	kJ/mol	Joback Method
hvap	70.61	kJ/mol	Joback Method
log10ws	-4.69		Crippen Method
logp	2.843		Crippen Method
mcvol	226.720	ml/mol	McGowan Method
pc	1678.28	kPa	Joback Method
rinpol	1575.00		NIST Webbook
rinpol	1530.00		NIST Webbook
rinpol	1485.00		NIST Webbook
rinpol	1530.00		NIST Webbook
rinpol	1575.00		NIST Webbook
tb	796.72	K	Joback Method
tc	983.13	K	Joback Method
tf	558.48	K	Joback Method
vc	0.929	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	680.34	J/molxK	796.72	Joback Method
cpg	689.23	J/molxK	827.79	Joback Method
cpg	697.35	J/molxK	858.86	Joback Method
cpg	704.76	J/molxK	889.93	Joback Method

cpg	711.50	J/mol×K	920.99	Joback Method
cpg	717.63	J/mol×K	952.06	Joback Method
cpg	723.19	J/mol×K	983.13	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375507&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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