

# Diglycolic acid, 2,4-dichloronaphth-1-yl ethyl ester

Inchi:	InChI=1S/C16H14Cl2O5/c1-2-22-14(19)8-21-9-15(20)23-16-11-6-4-3-5-10(11)12(17)7-13
InchiKey:	RNRCTXWRRNCISG-UHFFFAOYSA-N
Formula:	C16H14Cl2O5
SMILES:	CCOC(=O)COCC(=O)Oc1c(Cl)cc(Cl)c2ccccc12
Mol. weight [g/mol]:	357.19

## Physical Properties

Property code	Value	Unit	Source
gf	-322.69	kJ/mol	Joback Method
hf	-633.68	kJ/mol	Joback Method
hfus	42.24	kJ/mol	Joback Method
hvap	86.60	kJ/mol	Joback Method
log10ws	-4.58		Crippen Method
logp	3.632		Crippen Method
mcvol	238.310	ml/mol	McGowan Method
pc	2025.41	kPa	Joback Method
rinpola	3178.00		NIST Webbook
rinpola	3178.00		NIST Webbook
tb	875.94	K	Joback Method
tc	1103.53	K	Joback Method
tf	593.15	K	Joback Method
vc	0.909	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	645.87	J/molxK	875.94	Joback Method
cpg	688.28	J/molxK	1065.60	Joback Method
cpg	681.79	J/molxK	1027.67	Joback Method
cpg	674.31	J/molxK	989.73	Joback Method
cpg	665.85	J/molxK	951.80	Joback Method
cpg	656.37	J/molxK	913.87	Joback Method
cpg	693.81	J/molxK	1103.53	Joback Method
dvisc	0.0001109	Paxs	875.94	Joback Method

dvisc	0.0001313	Paxs	828.81	Joback Method
dvisc	0.0001586	Paxs	781.68	Joback Method
dvisc	0.0001963	Paxs	734.55	Joback Method
dvisc	0.0002503	Paxs	687.41	Joback Method
dvisc	0.0003307	Paxs	640.28	Joback Method
dvisc	0.0004566	Paxs	593.15	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382628&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382628&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/117-676-2/Diglycolic-acid-2-4-dichloronaphth-1-yl-ethyl-ester.pdf>

Generated by Cheméo on 2024-05-03 19:27:01.042655758 +0000 UTC m=+17053669.963233071.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.