

# 1,1'-Biphenyl, 4,4'-dinitro-

<b>Other names:</b>	Biphenyl, 4,4'-dinitro- 4,4'-Dinitrobiphenyl 4,4'-Dinitrodiphenyl 4,4'-Dinitrobifenyyl
<b>Inchi:</b>	InChI=1S/C12H8N2O4/c15-13(16)11-5-1-9(2-6-11)10-3-7-12(8-4-10)14(17)18/h1-8H
<b>InchiKey:</b>	BDLNCFCZHNKBGI-UHFFFAOYSA-N
<b>Formula:</b>	C12H8N2O4
<b>SMILES:</b>	O=[N+]([O-])c1ccc(-c2ccc([N+](=O)[O-])cc2)cc1
<b>Mol. weight [g/mol]:</b>	244.20
<b>CAS:</b>	1528-74-1

## Physical Properties

Property code	Value	Unit	Source
gf	326.82	kJ/mol	Joback Method
hf	137.59	kJ/mol	Joback Method
hfus	36.86	kJ/mol	Joback Method
hvap	81.36	kJ/mol	Joback Method
log10ws	-5.38		Crippen Method
logp	3.170		Crippen Method
mcvol	167.260	ml/mol	McGowan Method
pc	3403.91	kPa	Joback Method
tb	840.96	K	Joback Method
tc	1132.36	K	Joback Method
tf	590.10	K	Joback Method
vc	0.655	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	457.48	J/molxK	840.96	Joback Method
cpg	468.03	J/molxK	889.53	Joback Method
cpg	477.38	J/molxK	938.09	Joback Method
cpg	485.65	J/molxK	986.66	Joback Method
cpg	492.96	J/molxK	1035.23	Joback Method

cpg	499.43	J/mol×K	1083.80	Joback Method
cpg	505.17	J/mol×K	1132.36	Joback Method
hsubt	104.60 ± 1.80	kJ/mol	434.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1528741&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1528741&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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