

Dipropylene glycol mono-tert-butyl ether, pentafluoropropionate

Inchi: InChI=1S/C13H21F5O4/c1-8(7-21-11(3,4)5)20-6-9(2)22-10(19)12(14,15)13(16,17)18/h8-20
InchiKey: MRPIUAXKMRRSRV-UHFFFAOYSA-N
Formula: C13H21F5O4
SMILES: CC(COC(C)(C)C)OCC(C)OC(=O)C(F)(F)C(F)(F)F
Mol. weight [g/mol]: 336.30

Physical Properties

Property code	Value	Unit	Source
gf	-1355.75	kJ/mol	Joback Method
hf	-1838.25	kJ/mol	Joback Method
hfus	20.70	kJ/mol	Joback Method
hvap	49.76	kJ/mol	Joback Method
log10ws	-3.61		Crippen Method
logp	3.336		Crippen Method
mcvol	222.060	ml/mol	McGowan Method
pc	1472.49	kPa	Joback Method
rinpol	1169.00		NIST Webbook
rinpol	1169.00		NIST Webbook
tb	603.75	K	Joback Method
tc	768.78	K	Joback Method
tf	333.10	K	Joback Method
vc	0.869	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	621.63	J/mol×K	603.75	Joback Method
cpg	636.97	J/mol×K	631.25	Joback Method
cpg	651.50	J/mol×K	658.76	Joback Method
cpg	665.25	J/mol×K	686.26	Joback Method
cpg	678.25	J/mol×K	713.77	Joback Method
cpg	690.52	J/mol×K	741.27	Joback Method
cpg	702.09	J/mol×K	768.78	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U378342&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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