

# Succinic acid, 2,3-dichlorophenyl but-3-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C14H14Cl2O4/c1-2-3-9-19-12(17)7-8-13(18)20-11-6-4-5-10(15)14(11)16/h2,4-
<b>InchiKey:</b>	WMVFZNMVGGKIPZ-UHFFFAOYSA-N
<b>Formula:</b>	C14H14Cl2O4
<b>SMILES:</b>	C=CCCOC(=O)CCC(=O)Oc1cccc(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	317.17

## Physical Properties

Property code	Value	Unit	Source
gf	-243.71	kJ/mol	Joback Method
hf	-514.35	kJ/mol	Joback Method
hfus	37.97	kJ/mol	Joback Method
hvap	76.77	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	3.798		Crippen Method
mvol	219.420	ml/mol	McGowan Method
pc	2064.24	kPa	Joback Method
rinpol	2212.00		NIST Webbook
rinpol	2212.00		NIST Webbook
tb	780.48	K	Joback Method
tc	997.13	K	Joback Method
tf	501.40	K	Joback Method
vc	0.839	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	563.46	J/molxK	780.48	Joback Method
cpg	575.01	J/molxK	816.59	Joback Method
cpg	585.67	J/molxK	852.70	Joback Method
cpg	595.44	J/molxK	888.80	Joback Method
cpg	604.33	J/molxK	924.91	Joback Method
cpg	612.36	J/molxK	961.02	Joback Method
cpg	619.53	J/molxK	997.13	Joback Method
dvisc	0.0006383	Paxs	501.40	Joback Method

dvisc	0.0004128	Paxs	547.91	Joback Method
dvisc	0.0002859	Paxs	594.43	Joback Method
dvisc	0.0002088	Paxs	640.94	Joback Method
dvisc	0.0001591	Paxs	687.45	Joback Method
dvisc	0.0001255	Paxs	733.97	Joback Method
dvisc	0.0001018	Paxs	780.48	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391200&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391200&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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