

# Succinic acid, 2,2,3,3,4,4,5,5-octafluoropentyl geranyl ester

<b>Inchi:</b>	InChI=1S/C19H24F8O4/c1-12(2)5-4-6-13(3)9-10-30-14(28)7-8-15(29)31-11-17(22,23)19
<b>InchiKey:</b>	CPXDHJXBIGOBPU-UKTHLTGXSA-N
<b>Formula:</b>	C19H24F8O4
<b>SMILES:</b>	CC(C)=CCCC(C)=CCOC(=O)CCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
<b>Mol. weight [g/mol]:</b>	468.38

## Physical Properties

Property code	Value	Unit	Source
gf	-1767.80	kJ/mol	Joback Method
hf	-2310.64	kJ/mol	Joback Method
hfus	47.20	kJ/mol	Joback Method
hvap	65.46	kJ/mol	Joback Method
log10ws	-6.46		Crippen Method
logp	5.717		Crippen Method
mcvol	299.010	ml/mol	McGowan Method
pc	1020.08	kPa	Joback Method
rinpola	1995.00		NIST Webbook
rinpola	1995.00		NIST Webbook
tb	778.81	K	Joback Method
tc	955.62	K	Joback Method
tf	407.11	K	Joback Method
vc	1.214	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	923.73	J/mol×K	778.81	Joback Method
cpg	938.18	J/mol×K	808.28	Joback Method
cpg	951.77	J/mol×K	837.75	Joback Method
cpg	964.58	J/mol×K	867.21	Joback Method
cpg	976.67	J/mol×K	896.68	Joback Method
cpg	988.12	J/mol×K	926.15	Joback Method
cpg	998.98	J/mol×K	955.62	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391209&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391209&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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