

# Butanamide, N-(4-chlorophenyl)-3-oxo-

**Other names:**

Acetoacetanilide, 4'-chloro-  
p-Chloroacetoacetanilide  
Acetoacet-p-chloroanilide  
Acetoacetyl-4-chloroanilide  
4'-Chloroacetoacetanilide  
Acetoacetic acid-4-chloroanilide  
Acetoacetanilide, p-chloro-  
N-(4-Chlorophenyl)acetoacetamide  
Butanamide, 3-oxo-N-(4-chlorophenyl)-  
N-(4-Chloro-phenyl)-3-oxo-butyramide  
N-(4-Chlorophenyl)acetylacetamide  
NSC 3544

**Inchi:**

InChI=1S/C10H10ClNO2/c1-7(13)6-10(14)12-9-4-2-8(11)3-5-9/h2-5H,6H2,1H3,(H,12,14)

**InchiKey:**

JMRJWEJJUKUBEA-UHFFFAOYSA-N

**Formula:**

C10H10ClNO2

**SMILES:**

CC(=O)CC(=O)Nc1ccc(Cl)cc1

**Mol. weight [g/mol]:**

211.65

**CAS:**

101-92-8

## Physical Properties

Property code	Value	Unit	Source
gf	-44.28	kJ/mol	Joback Method
hf	-212.10	kJ/mol	Joback Method
hfus	27.80	kJ/mol	Joback Method
hvap	65.11	kJ/mol	Joback Method
log10ws	-2.48		Crippen Method
logp	2.258		Crippen Method
mcvol	153.360	ml/mol	McGowan Method
pc	3228.31	kPa	Joback Method
tb	655.20	K	Joback Method
tc	882.11	K	Joback Method
tf	423.84	K	Joback Method
vc	0.584	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	361.94	J/mol×K	655.20	Joback Method
cpg	373.15	J/mol×K	693.02	Joback Method
cpg	383.55	J/mol×K	730.84	Joback Method
cpg	393.15	J/mol×K	768.65	Joback Method
cpg	402.00	J/mol×K	806.47	Joback Method
cpg	410.14	J/mol×K	844.29	Joback Method
cpg	417.60	J/mol×K	882.11	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C101928&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C101928&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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