

Fumaric acid, 2,4,4-trimethylpentyl 2,2,3,3-tetrafluoropropyl ester

Inchi: InChI=1S/C15H22F4O4/c1-10(7-14(2,3)4)8-22-11(20)5-6-12(21)23-9-15(18,19)13(16)17
InchiKey: VCBPVEHYIFBMGW-AATRIKPKSA-N
Formula: C15H22F4O4
SMILES: CC(COC(=O)C=CC(=O)OCC(F)(F)C(F)F)CC(C)(C)C
Mol. weight [g/mol]: 342.33

Physical Properties

Property code	Value	Unit	Source
gf	-1090.64	kJ/mol	Joback Method
hf	-1537.81	kJ/mol	Joback Method
hfus	30.83	kJ/mol	Joback Method
hvap	60.62	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.602		Crippen Method
mvol	239.870	ml/mol	McGowan Method
pc	1434.80	kPa	Joback Method
rinpol	1620.00		NIST Webbook
rinpol	1620.00		NIST Webbook
tb	689.08	K	Joback Method
tc	865.41	K	Joback Method
tf	375.25	K	Joback Method
vc	0.942	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	691.98	J/molxK	689.08	Joback Method
cpg	706.65	J/molxK	718.47	Joback Method
cpg	720.47	J/molxK	747.86	Joback Method
cpg	733.49	J/molxK	777.25	Joback Method
cpg	745.74	J/molxK	806.64	Joback Method
cpg	757.26	J/molxK	836.03	Joback Method
cpg	768.09	J/molxK	865.41	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U405598&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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