

L-Phenylalanine, N-caproyl-, methyl ester

Inchi:	InChI=1S/C16H23NO3/c1-3-4-6-11-15(18)17-14(16(19)20-2)12-13-9-7-5-8-10-13/h5,7-10
InchiKey:	IZQJDSLVCFGOEO-UHFFFAOYSA-N
Formula:	C16H23NO3
SMILES:	CCCCC(=O)NC(Cc1ccccc1)C(=O)OC
Mol. weight [g/mol]:	277.36

Physical Properties

Property code	Value	Unit	Source
gf	-79.64	kJ/mol	Joback Method
hf	-446.23	kJ/mol	Joback Method
hfus	37.20	kJ/mol	Joback Method
hvap	75.44	kJ/mol	Joback Method
log10ws	-3.56		Crippen Method
logp	2.467		Crippen Method
mvol	231.530	ml/mol	McGowan Method
pc	1901.92	kPa	Joback Method
rinpol	2023.00		NIST Webbook
rinpol	2023.00		NIST Webbook
tb	772.05	K	Joback Method
tc	978.12	K	Joback Method
tf	456.25	K	Joback Method
vc	0.882	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	678.31	J/mol×K	772.05	Joback Method
cpg	693.41	J/mol×K	806.40	Joback Method
cpg	707.49	J/mol×K	840.74	Joback Method
cpg	720.56	J/mol×K	875.09	Joback Method
cpg	732.67	J/mol×K	909.43	Joback Method
cpg	743.85	J/mol×K	943.78	Joback Method
cpg	754.13	J/mol×K	978.12	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U299737&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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