

# Sulphenone

<b>Other names:</b>	Benzene, 1-chloro-4-(phenylsulfonyl)- Sulfone, p-chlorophenyl phenyl p-Chlorophenyl phenyl sulfone R 242 Sulfenon 4-Chlorodiphenyl sulfone 4-Chlorophenyl phenyl sulfone p-Monochlorophenyl phenyl sulfone R-242-B Compound R-242 ENT 17,941 Sulfenone Trifenson 4-Chlordifenylsulfon 4-Chlorodiphenyl sulphone p-Chlorophenyl phenyl sulphone 1-Chloro-4-(phenylsulfonyl)benzene NSC 404335
<b>Inchi:</b>	InChI=1S/C12H9ClO2S/c13-10-6-8-12(9-7-10)16(14,15)11-4-2-1-3-5-11/h1-9H
<b>InchiKey:</b>	OF CFY WOKHPOXKF-UHFFFAOYSA-N
<b>Formula:</b>	C12H9ClO2S
<b>SMILES:</b>	O=S(=O)(c1ccccc1)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	252.72
<b>CAS:</b>	80-00-2

## Physical Properties

Property code	Value	Unit	Source
gf	-215.12	kJ/mol	Joback Method
hf	-298.51	kJ/mol	Joback Method
hfus	30.10	kJ/mol	Joback Method
hvap	70.54	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.173		Crippen Method
mcvol	172.750	ml/mol	McGowan Method
pc	3881.95	kPa	Joback Method
tb	617.51	K	Joback Method
tc	860.39	K	Joback Method

tf	358.84	K	Joback Method
vc	0.666	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	385.19	J/mol×K	617.51	Joback Method
cpg	399.54	J/mol×K	657.99	Joback Method
cpg	412.66	J/mol×K	698.47	Joback Method
cpg	424.60	J/mol×K	738.95	Joback Method
cpg	435.39	J/mol×K	779.43	Joback Method
cpg	445.07	J/mol×K	819.91	Joback Method
cpg	453.69	J/mol×K	860.39	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C80002&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C80002&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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