

# Diglycolic acid, isobutyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C14H15Cl3O5/c1-8(2)5-21-12(18)6-20-7-13(19)22-14-10(16)3-9(15)4-11(14)17
InchiKey:	NVYHZCGKBVPWNX-UHFFFAOYSA-N
Formula:	C14H15Cl3O5
SMILES:	CC(C)COC(=O)COCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	369.62

## Physical Properties

Property code	Value	Unit	Source
gf	-460.55	kJ/mol	Joback Method
hf	-804.49	kJ/mol	Joback Method
hfus	40.72	kJ/mol	Joback Method
hvap	84.51	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.768		Crippen Method
mcvol	241.830	ml/mol	McGowan Method
pc	1883.80	kPa	Joback Method
rinpol	2812.00		NIST Webbook
rinpol	2812.00		NIST Webbook
tb	848.19	K	Joback Method
tc	1067.72	K	Joback Method
tf	552.83	K	Joback Method
vc	0.918	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	635.60	J/molxK	848.19	Joback Method
cpg	646.25	J/molxK	884.78	Joback Method
cpg	655.85	J/molxK	921.37	Joback Method
cpg	664.38	J/molxK	957.95	Joback Method
cpg	671.83	J/molxK	994.54	Joback Method
cpg	678.17	J/molxK	1031.13	Joback Method
cpg	683.40	J/molxK	1067.72	Joback Method
dvisc	0.0003656	Paxs	552.83	Joback Method

dvisc	0.0002375	Paxs	602.06	Joback Method
dvisc	0.0001646	Paxs	651.28	Joback Method
dvisc	0.0001201	Paxs	700.51	Joback Method
dvisc	0.0000914	Paxs	749.74	Joback Method
dvisc	0.0000719	Paxs	798.96	Joback Method
dvisc	0.0000582	Paxs	848.19	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382733&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382733&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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