

# Diglycolic acid, butyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C14H15Cl3O5/c1-2-3-4-21-12(18)7-20-8-13(19)22-14-10(16)5-9(15)6-11(14)17
InchiKey:	ADAKUVKETHXTSS-UHFFFAOYSA-N
Formula:	C14H15Cl3O5
SMILES:	CCCCOC(=O)COCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	369.62

## Physical Properties

Property code	Value	Unit	Source
gf	-458.11	kJ/mol	Joback Method
hf	-799.21	kJ/mol	Joback Method
hfus	44.24	kJ/mol	Joback Method
hvap	84.90	kJ/mol	Joback Method
log10ws	-4.30		Crippen Method
logp	3.912		Crippen Method
mvol	241.830	ml/mol	McGowan Method
pc	1870.79	kPa	Joback Method
rinpol	2874.00		NIST Webbook
rinpol	2874.00		NIST Webbook
tb	848.63	K	Joback Method
tc	1065.41	K	Joback Method
tf	567.83	K	Joback Method
vc	0.924	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	635.00	J/molxK	848.63	Joback Method
cpg	677.40	J/molxK	1029.28	Joback Method
cpg	671.01	J/molxK	993.15	Joback Method
cpg	663.56	J/molxK	957.02	Joback Method
cpg	655.07	J/molxK	920.89	Joback Method
cpg	645.54	J/molxK	884.76	Joback Method
cpg	682.72	J/molxK	1065.41	Joback Method
dvisc	0.0000635	Paxs	848.63	Joback Method

dvisc	0.0000774	Paxs	801.83	Joback Method
dvisc	0.0000967	Paxs	755.03	Joback Method
dvisc	0.0001244	Paxs	708.23	Joback Method
dvisc	0.0001659	Paxs	661.43	Joback Method
dvisc	0.0002312	Paxs	614.63	Joback Method
dvisc	0.0003402	Paxs	567.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U382734&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U382734&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/118-317-9/Diglycolic-acid-butyl-2-4-6-trichlorophenyl-ester.pdf>

Generated by Cheméo on 2024-05-02 04:35:05.724354535 +0000 UTC m=+16913754.644931873.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.