

# Urea, trimethylnitroso-

<b>Other names:</b>	N-Nitrosotrimethylurea N-Trimethylnitrosourea Nitrosotrimethylurea Trimethylnitrosourea Urea, 1,1,3-trimethyl-3-nitroso- N-Nitroso-trimethylharnstoff Trimethylnitrosoharnstoff Trimethylnitrosomocovina N-Trimethyl-N-nitrosourea 1,1,3-Trimethyl-3-nitrosourea Urea, N,N,N'-trimethyl-N'-nitroso-
<b>Inchi:</b>	InChI=1S/C4H9N3O2/c1-6(2)4(8)7(3)5-9/h1-3H3
<b>InchiKey:</b>	LOEHVDVYQTQIWEV-UHFFFAOYSA-N
<b>Formula:</b>	C4H9N3O2
<b>SMILES:</b>	CN(C)C(=O)N(C)N=O
<b>Mol. weight [g/mol]:</b>	131.13
<b>CAS:</b>	3475-63-6

## Physical Properties

Property code	Value	Unit	Source
hf	-271.60	kJ/mol	Joback Method
hvap	44.43	kJ/mol	Joback Method
log10ws	-0.57		Crippen Method
logp	0.281		Crippen Method
mcvol	100.300	ml/mol	McGowan Method
pc	4103.88	kPa	Joback Method
tb	433.07	K	Joback Method
tc	610.97	K	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3475636&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3475636&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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