

# Glutaric acid, 2,2-dichloroethyl tridecyl ester

**Inchi:** InChI=1S/C20H36Cl2O4/c1-2-3-4-5-6-7-8-9-10-11-12-16-25-19(23)14-13-15-20(24)26-17  
**InchiKey:** OXSYLOSHLHSLLI-UHFFFAOYSA-N  
**Formula:** C20H36Cl2O4  
**SMILES:** CCCCCCCCCCCCCOC(=O)CCCC(=O)OCC(Cl)Cl  
**Mol. weight [g/mol]:** 411.40

## Physical Properties

Property code	Value	Unit	Source
gf	-376.62	kJ/mol	Joback Method
hf	-982.49	kJ/mol	Joback Method
hfus	58.00	kJ/mol	Joback Method
hvap	86.81	kJ/mol	Joback Method
log10ws	-6.83		Crippen Method
logp	6.358		Crippen Method
mcvol	332.020	ml/mol	McGowan Method
pc	1032.57	kPa	Joback Method
rinpola	2734.00		NIST Webbook
rinpola	2734.00		NIST Webbook
tb	884.00	K	Joback Method
tc	1082.74	K	Joback Method
tf	504.32	K	Joback Method
vc	1.296	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1030.76	J/molxK	884.00	Joback Method
cpg	1047.19	J/molxK	917.12	Joback Method
cpg	1062.45	J/molxK	950.25	Joback Method
cpg	1076.59	J/molxK	983.37	Joback Method
cpg	1089.62	J/molxK	1016.50	Joback Method
cpg	1101.56	J/molxK	1049.62	Joback Method
cpg	1112.45	J/molxK	1082.74	Joback Method
dvisc	0.0005909	Paxs	504.32	Joback Method

dvisc	0.0002842	Paxs	567.60	Joback Method
dvisc	0.0001583	Paxs	630.88	Joback Method
dvisc	0.0000981	Paxs	694.16	Joback Method
dvisc	0.0000659	Paxs	757.44	Joback Method
dvisc	0.0000470	Paxs	820.72	Joback Method
dvisc	0.0000352	Paxs	884.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391628&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391628&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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