

# 5-Phenylvaleric acid, 4-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C17H17ClO2/c18-15-10-12-16(13-11-15)20-17(19)9-5-4-8-14-6-2-1-3-7-14/h1-
<b>InchiKey:</b>	QTDNOYVRUFYGHX-UHFFFAOYSA-N
<b>Formula:</b>	C17H17ClO2
<b>SMILES:</b>	O=C(CCCCc1ccccc1)Oc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	288.77

## Physical Properties

Property code	Value	Unit	Source
gf	61.60	kJ/mol	Joback Method
hf	-193.16	kJ/mol	Joback Method
hfus	34.46	kJ/mol	Joback Method
hvap	72.19	kJ/mol	Joback Method
log10ws	-5.34		Crippen Method
logp	4.658		Crippen Method
mvol	222.550	ml/mol	McGowan Method
pc	2077.43	kPa	Joback Method
rinpol	2272.00		NIST Webbook
rinpol	2272.00		NIST Webbook
tb	760.42	K	Joback Method
tc	992.42	K	Joback Method
tf	448.79	K	Joback Method
vc	0.845	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	597.51	J/molxK	760.42	Joback Method
cpg	612.43	J/molxK	799.09	Joback Method
cpg	626.16	J/molxK	837.75	Joback Method
cpg	638.76	J/molxK	876.42	Joback Method
cpg	650.28	J/molxK	915.09	Joback Method
cpg	660.78	J/molxK	953.76	Joback Method
cpg	670.31	J/molxK	992.42	Joback Method
dvisc	0.0009408	Paxs	448.79	Joback Method

dvisc	0.0005320	Paxs	500.73	Joback Method
dvisc	0.0003349	Paxs	552.67	Joback Method
dvisc	0.0002282	Paxs	604.61	Joback Method
dvisc	0.0001653	Paxs	656.54	Joback Method
dvisc	0.0001255	Paxs	708.48	Joback Method
dvisc	0.0000989	Paxs	760.42	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406900&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406900&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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