

# Glutaric acid, 2,3-dichlorophenyl 2,4-dimethylpent-3-yl ester

**Inchi:** InChI=1S/C18H24Cl2O4/c1-11(2)18(12(3)4)24-16(22)10-6-9-15(21)23-14-8-5-7-13(19)17  
**InchiKey:** HRCFEXPYMZABAK-UHFFFAOYSA-N  
**Formula:** C18H24Cl2O4  
**SMILES:** CC(C)C(OC(=O)CCCC(=O)Oc1cccc(Cl)c1Cl)C(C)C  
**Mol. weight [g/mol]:** 375.29

## Physical Properties

Property code	Value	Unit	Source
gf	-305.19	kJ/mol	Joback Method
hf	-738.18	kJ/mol	Joback Method
hfus	39.04	kJ/mol	Joback Method
hvap	85.18	kJ/mol	Joback Method
log10ws	-5.83		Crippen Method
logp	5.293		Crippen Method
mcvol	280.080	ml/mol	McGowan Method
pc	1475.88	kPa	Joback Method
rinpol	2477.00		NIST Webbook
rinpol	2477.00		NIST Webbook
tb	874.00	K	Joback Method
tc	1089.91	K	Joback Method
tf	503.24	K	Joback Method
vc	1.063	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	814.09	J/molxK	874.00	Joback Method
cpg	827.69	J/molxK	909.99	Joback Method
cpg	840.11	J/molxK	945.97	Joback Method
cpg	851.35	J/molxK	981.96	Joback Method
cpg	861.44	J/molxK	1017.94	Joback Method
cpg	870.40	J/molxK	1053.93	Joback Method
cpg	878.25	J/molxK	1089.91	Joback Method
dvisc	0.0005880	Paxs	503.24	Joback Method

dvisc	0.0002979	Paxs	565.03	Joback Method
dvisc	0.0001726	Paxs	626.83	Joback Method
dvisc	0.0001103	Paxs	688.62	Joback Method
dvisc	0.0000758	Paxs	750.41	Joback Method
dvisc	0.0000552	Paxs	812.21	Joback Method
dvisc	0.0000421	Paxs	874.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393483&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393483&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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