

# Adipic acid, decyl 2,2,3,3,4,4,5,5-octafluoropentyl ester

<b>Inchi:</b>	InChI=1S/C21H32F8O4/c1-2-3-4-5-6-7-8-11-14-32-16(30)12-9-10-13-17(31)33-15-19(24
<b>InchiKey:</b>	MHLDTAZGCARVKW-UHFFFAOYSA-N
<b>Formula:</b>	C21H32F8O4
<b>SMILES:</b>	CCCCCCCCCOC(=O)CCCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
<b>Mol. weight [g/mol]:</b>	500.46

## Physical Properties

Property code	Value	Unit	Source
gf	-1894.30	kJ/mol	Joback Method
hf	-2566.78	kJ/mol	Joback Method
hfus	54.59	kJ/mol	Joback Method
hvap	69.84	kJ/mol	Joback Method
log10ws	-7.59		Crippen Method
logp	6.945		Crippen Method
mcvol	335.790	ml/mol	McGowan Method
pc	846.53	kPa	Joback Method
rinpol	2211.00		NIST Webbook
rinpol	2211.00		NIST Webbook
tb	816.49	K	Joback Method
tc	1000.80	K	Joback Method
tf	467.73	K	Joback Method
vc	1.365	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1092.24	J/molxK	816.49	Joback Method
cpg	1109.25	J/molxK	847.21	Joback Method
cpg	1125.20	J/molxK	877.93	Joback Method
cpg	1140.14	J/molxK	908.64	Joback Method
cpg	1154.15	J/molxK	939.36	Joback Method
cpg	1167.28	J/molxK	970.08	Joback Method
cpg	1179.59	J/molxK	1000.80	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353736&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353736&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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