

# Ethanediamide, N,N'-diphenyl-

**Other names:**

Oxanilide  
N,N'-Diphenyloxamide  
Oxalanilide  
Oxamide, N,N'-diphenyl-  
Oxanilid  
Oxaldianilide  
N,N'-Diphenyloxalamide  
Bis-(phenyl-carbamyl)  
Ethanediamide, N1,N2-diphenyl-  
NSC 4183

**Inchi:**

InChI=1S/C14H12N2O2/c17-13(15-11-7-3-1-4-8-11)14(18)16-12-9-5-2-6-10-12/h1-10H,(

**InchiKey:**

FTWUXYZHDFCGSV-UHFFFAOYSA-N

**Formula:**

C14H12N2O2

**SMILES:**

O=C(Nc1ccccc1)C(=O)Nc1ccccc1

**Mol. weight [g/mol]:**

240.26

**CAS:**

620-81-5

## Physical Properties

| Property code | Value   | Unit                 | Source         |
|---------------|---------|----------------------|----------------|
| gf            | 212.76  | kJ/mol               | Joback Method  |
| hf            | 22.55   | kJ/mol               | Joback Method  |
| hfus          | 33.49   | kJ/mol               | Joback Method  |
| hvap          | 77.67   | kJ/mol               | Joback Method  |
| log10ws       | -2.69   |                      | Crippen Method |
| logp          | 2.264   |                      | Crippen Method |
| mcvol         | 183.700 | ml/mol               | McGowan Method |
| pc            | 3272.78 | kPa                  | Joback Method  |
| tb            | 781.16  | K                    | Joback Method  |
| tc            | 1026.92 | K                    | Joback Method  |
| tf            | 505.56  | K                    | Joback Method  |
| vc            | 0.685   | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 503.59 | J/mol×K | 781.16          | Joback Method |
| cpg           | 515.77 | J/mol×K | 822.12          | Joback Method |
| cpg           | 526.79 | J/mol×K | 863.08          | Joback Method |
| cpg           | 536.74 | J/mol×K | 904.04          | Joback Method |
| cpg           | 545.71 | J/mol×K | 945.00          | Joback Method |
| cpg           | 553.79 | J/mol×K | 985.96          | Joback Method |
| cpg           | 561.06 | J/mol×K | 1026.92         | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C620815&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C620815&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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