

Terephthalic acid, nonyl 2,2,3,3,3-pentafluoropropyl ester

Inchi:	InChI=1S/C20H25F5O4/c1-2-3-4-5-6-7-8-13-28-17(26)15-9-11-16(12-10-15)18(27)29-14
InchiKey:	VOMZPKYFHCASNM-UHFFFAOYSA-N
Formula:	C20H25F5O4
SMILES:	CCCCCCCCCOC(=O)c1ccc(C(=O)OCC(F)(F)C(F)(F)F)cc1
Mol. weight [g/mol]:	424.40

Physical Properties

Property code	Value	Unit	Source
gf	-1215.91	kJ/mol	Joback Method
hf	-1718.72	kJ/mol	Joback Method
hfus	47.35	kJ/mol	Joback Method
hvap	74.69	kJ/mol	Joback Method
log10ws	-7.12		Crippen Method
logp	5.948		Crippen Method
mvol	292.630	ml/mol	McGowan Method
pc	1175.24	kPa	Joback Method
rinpol	2278.00		NIST Webbook
rinpol	2278.00		NIST Webbook
tb	831.13	K	Joback Method
tc	1021.80	K	Joback Method
tf	506.21	K	Joback Method
vc	1.163	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	916.89	J/mol×K	831.13	Joback Method
cpg	931.33	J/mol×K	862.91	Joback Method
cpg	944.78	J/mol×K	894.69	Joback Method
cpg	957.28	J/mol×K	926.47	Joback Method
cpg	968.90	J/mol×K	958.25	Joback Method
cpg	979.69	J/mol×K	990.02	Joback Method
cpg	989.69	J/mol×K	1021.80	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415780&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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