

# 6-Fluoro-2-trifluoromethylbenzoic acid,2,4-dichloronaphthyl-1 ester

**Other names:** 6-Fluoro-2-trifluorobenzoic acid, 2,4-dichloronaphthyl-1 ester

**Inchi:** InChI=1S/C18H8Cl2F4O2/c19-12-8-13(20)16(10-5-2-1-4-9(10)12)26-17(25)15-11(18(22,

**InchiKey:** VIIOZFFIPGPZEZ-UHFFFAOYSA-N

**Formula:** C18H8Cl2F4O2

**SMILES:** O=C(Oc1c(Cl)cc(Cl)c2cccc12)c1c(F)cccc1C(F)(F)F

**Mol. weight [g/mol]:** 403.15

## Physical Properties

Property code	Value	Unit	Source
gf	-650.18	kJ/mol	Joback Method
hf	-877.54	kJ/mol	Joback Method
hfus	41.62	kJ/mol	Joback Method
hvap	78.53	kJ/mol	Joback Method
log10ws	-8.17		Crippen Method
logp	6.524		Crippen Method
mcvol	236.500	ml/mol	McGowan Method
pc	1885.44	kPa	Joback Method
rinpol	2437.00		NIST Webbook
rinpol	2437.00		NIST Webbook
tb	853.48	K	Joback Method
tc	1085.89	K	Joback Method
tf	577.54	K	Joback Method
vc	0.932	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	625.74	J/molxK	853.48	Joback Method
cpg	635.53	J/molxK	892.22	Joback Method
cpg	644.51	J/molxK	930.95	Joback Method
cpg	652.75	J/molxK	969.69	Joback Method
cpg	660.34	J/molxK	1008.42	Joback Method
cpg	667.39	J/molxK	1047.16	Joback Method
cpg	673.97	J/molxK	1085.89	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U343747&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U343747&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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