

# Glutaric acid, di(4-nitrophenyl) ester

**Inchi:** InChI=1S/C17H14N2O8/c20-16(26-14-8-4-12(5-9-14)18(22)23)2-1-3-17(21)27-15-10-6-1  
**InchiKey:** IOIJXIHQQRSQDS-UHFFFAOYSA-N  
**Formula:** C17H14N2O8  
**SMILES:** O=C(CCCC(=O)Oc1ccc([N+](=O)[O-])cc1)Oc1ccc([N+](=O)[O-])cc1  
**Mol. weight [g/mol]:** 374.30

## Physical Properties

| Property code | Value   | Unit                 | Source         |
|---------------|---------|----------------------|----------------|
| gf            | -98.92  | kJ/mol               | Joback Method  |
| hf            | -455.21 | kJ/mol               | Joback Method  |
| hfus          | 55.39   | kJ/mol               | Joback Method  |
| hvap          | 110.81  | kJ/mol               | Joback Method  |
| log10ws       | -5.47   |                      | Crippen Method |
| logp          | 3.184   |                      | Crippen Method |
| mcvol         | 252.590 | ml/mol               | McGowan Method |
| pc            | 2263.26 | kPa                  | Joback Method  |
| rinpol        | 3316.00 |                      | NIST Webbook   |
| rinpol        | 3316.00 |                      | NIST Webbook   |
| tb            | 1107.94 | K                    | Joback Method  |
| tc            | 1375.35 | K                    | Joback Method  |
| tf            | 790.77  | K                    | Joback Method  |
| vc            | 0.984   | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 779.53 | J/mol×K | 1107.94         | Joback Method |
| cpg           | 784.66 | J/mol×K | 1152.51         | Joback Method |
| cpg           | 788.33 | J/mol×K | 1197.08         | Joback Method |
| cpg           | 790.60 | J/mol×K | 1241.64         | Joback Method |
| cpg           | 791.53 | J/mol×K | 1286.21         | Joback Method |
| cpg           | 791.17 | J/mol×K | 1330.78         | Joback Method |
| cpg           | 789.57 | J/mol×K | 1375.35         | Joback Method |

# Sources

|                        |   |
|------------------------|---|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391981&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391981&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |

# Legend

|                  |   |
|------------------|---|
| <b>cpg:</b>      | Ideal gas heat capacity                         |
| <b>gf:</b>       | Standard Gibbs free energy of formation         |
| <b>hf:</b>       | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>     | Enthalpy of fusion at standard conditions       |
| <b>h vap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b>  | Log10 of Water solubility in mol/l              |
| <b>logp:</b>     | Octanol/Water partition coefficient             |
| <b>mcvol:</b>    | McGowan's characteristic volume                 |
| <b>pc:</b>       | Critical Pressure                               |
| <b>r in pol:</b> | Non-polar retention indices                     |
| <b>tb:</b>       | Normal Boiling Point Temperature                |
| <b>tc:</b>       | Critical Temperature                            |
| <b>tf:</b>       | Normal melting (fusion) point                   |
| <b>vc:</b>       | Critical Volume                                 |

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