

# Isophthalic acid, 2,6-dichlorophenyl ethyl ester

Inchi:	InChI=1S/C16H12Cl2O4/c1-2-21-15(19)10-5-3-6-11(9-10)16(20)22-14-12(17)7-4-8-13(14)
InchiKey:	ZAFPHHSXVIVZJU-UHFFFAOYSA-N
Formula:	C16H12Cl2O4
SMILES:	CCOC(=O)c1cccc(C(=O)Oc2c(Cl)cccc2Cl)c1
Mol. weight [g/mol]:	339.17

## Physical Properties

Property code	Value	Unit	Source
gf	-211.93	kJ/mol	Joback Method
hf	-456.00	kJ/mol	Joback Method
hfus	38.08	kJ/mol	Joback Method
hvap	84.83	kJ/mol	Joback Method
log10ws	-5.59		Crippen Method
logp	4.389		Crippen Method
mcvol	228.140	ml/mol	McGowan Method
pc	2227.09	kPa	Joback Method
rinpol	2534.00		NIST Webbook
rinpol	2534.00		NIST Webbook
tb	861.22	K	Joback Method
tc	1103.13	K	Joback Method
tf	564.64	K	Joback Method
vc	0.862	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	598.64	J/molxK	861.22	Joback Method
cpg	609.25	J/molxK	901.54	Joback Method
cpg	618.67	J/molxK	941.86	Joback Method
cpg	626.91	J/molxK	982.17	Joback Method
cpg	634.01	J/molxK	1022.49	Joback Method
cpg	639.97	J/molxK	1062.81	Joback Method
cpg	644.82	J/molxK	1103.13	Joback Method
dvisc	0.0004213	Paxs	564.64	Joback Method

dvisc	0.0002823	Paxs	614.07	Joback Method
dvisc	0.0002007	Paxs	663.50	Joback Method
dvisc	0.0001497	Paxs	712.93	Joback Method
dvisc	0.0001159	Paxs	762.36	Joback Method
dvisc	0.0000926	Paxs	811.79	Joback Method
dvisc	0.0000759	Paxs	861.22	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344618&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344618&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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