

Isophthalic acid, 2,6-dichlorophenyl propyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C17H14Cl2O4/c1-2-9-22-16(20)11-5-3-6-12(10-11)17(21)23-15-13(18)7-4-8-14 |
| InchiKey: | ANNVAZPXYKQEQR-UHFFFAOYSA-N |
| Formula: | C17H14Cl2O4 |
| SMILES: | CCCOC(=O)c1cccc(C(=O)Oc2c(Cl)cccc2Cl)c1 |
| Mol. weight [g/mol]: | 353.20 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | -203.51 | kJ/mol | Joback Method |
| hf | -476.64 | kJ/mol | Joback Method |
| hfus | 40.67 | kJ/mol | Joback Method |
| hvap | 87.06 | kJ/mol | Joback Method |
| log10ws | -6.01 | | Crippen Method |
| logp | 4.779 | | Crippen Method |
| mcvol | 242.230 | ml/mol | McGowan Method |
| pc | 2038.23 | kPa | Joback Method |
| rinpol | 2639.00 | | NIST Webbook |
| rinpol | 2639.00 | | NIST Webbook |
| tb | 884.10 | K | Joback Method |
| tc | 1122.62 | K | Joback Method |
| tf | 575.91 | K | Joback Method |
| vc | 0.917 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 653.59 | J/molxK | 884.10 | Joback Method |
| cpg | 695.80 | J/molxK | 1082.87 | Joback Method |
| cpg | 689.70 | J/molxK | 1043.11 | Joback Method |
| cpg | 682.45 | J/molxK | 1003.36 | Joback Method |
| cpg | 674.03 | J/molxK | 963.61 | Joback Method |
| cpg | 664.42 | J/molxK | 923.85 | Joback Method |
| cpg | 700.78 | J/molxK | 1122.62 | Joback Method |
| dvisc | 0.0000663 | Paxs | 884.10 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000811 | Paxs | 832.73 | Joback Method |
| dvisc | 0.0001020 | Paxs | 781.37 | Joback Method |
| dvisc | 0.0001325 | Paxs | 730.00 | Joback Method |
| dvisc | 0.0001791 | Paxs | 678.64 | Joback Method |
| dvisc | 0.0002542 | Paxs | 627.28 | Joback Method |
| dvisc | 0.0003841 | Paxs | 575.91 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U344619&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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