

2-Fluoro-3-trifluoromethylbenzoic acid, 3-hexadecyl ester

Inchi:	InChI=1S/C24H36F4O2/c1-3-5-6-7-8-9-10-11-12-13-14-16-19(4-2)30-23(29)20-17-15-18
InchiKey:	QHQWVHABXBKBRU-UHFFFAOYSA-N
Formula:	C24H36F4O2
SMILES:	CCCCCCCCCCCCC(CC)OC(=O)c1cccc(C(F)(F)F)c1F
Mol. weight [g/mol]:	432.54

Physical Properties

Property code	Value	Unit	Source
gf	-768.41	kJ/mol	Joback Method
hf	-1368.37	kJ/mol	Joback Method
hfus	55.35	kJ/mol	Joback Method
hvap	76.82	kJ/mol	Joback Method
log10ws	-9.54		Crippen Method
logp	8.481		Crippen Method
mcvol	339.780	ml/mol	McGowan Method
pc	902.89	kPa	Joback Method
rinpol	2446.00		NIST Webbook
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tb	854.86	K	Joback Method
tc	1047.04	K	Joback Method
tf	473.64	K	Joback Method
vc	1.351	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1097.37	J/mol×K	854.86	Joback Method
cpg	1115.45	J/mol×K	886.89	Joback Method
cpg	1132.40	J/mol×K	918.92	Joback Method
cpg	1148.30	J/mol×K	950.95	Joback Method
cpg	1163.19	J/mol×K	982.98	Joback Method
cpg	1177.13	J/mol×K	1015.01	Joback Method
cpg	1190.17	J/mol×K	1047.04	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U338528&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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