

# Chloromethyl 4-chlorodecanoate

<b>Other names:</b>	4-Chlorodecanoic acid, chloromethyl ester
<b>Inchi:</b>	InChI=1S/C11H20Cl2O2/c1-2-3-4-5-6-10(13)7-8-11(14)15-9-12/h10H,2-9H2,1H3
<b>InchiKey:</b>	LDNMSCDSUKAAMZ-UHFFFAOYSA-N
<b>Formula:</b>	C11H20Cl2O2
<b>SMILES:</b>	CCCCCCC(Cl)CCC(=O)OCCI
<b>Mol. weight [g/mol]:</b>	255.18
<b>CAS:</b>	80418-81-1

## Physical Properties

Property code	Value	Unit	Source
gf	-218.48	kJ/mol	Joback Method
hf	-551.93	kJ/mol	Joback Method
hfus	31.90	kJ/mol	Joback Method
hvap	57.62	kJ/mol	Joback Method
log10ws	-4.20		Crippen Method
logp	4.084		Crippen Method
mcvol	197.770	ml/mol	McGowan Method
pc	1900.26	kPa	Joback Method
rinpol	1654.00		NIST Webbook
rinpol	1658.00		NIST Webbook
rinpol	1650.00		NIST Webbook
rinpol	1650.00		NIST Webbook
rinpol	1664.00		NIST Webbook
ripol	2227.00		NIST Webbook
ripol	2193.00		NIST Webbook
ripol	2212.00		NIST Webbook
ripol	2193.00		NIST Webbook
ripol	2233.00		NIST Webbook
tb	601.79	K	Joback Method
tc	785.44	K	Joback Method
tf	330.73	K	Joback Method
vc	0.767	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	476.80	J/molxK	601.79	Joback Method
cpg	490.81	J/molxK	632.40	Joback Method
cpg	504.16	J/molxK	663.01	Joback Method
cpg	516.86	J/molxK	693.62	Joback Method
cpg	528.94	J/molxK	724.22	Joback Method
cpg	540.40	J/molxK	754.83	Joback Method
cpg	551.25	J/molxK	785.44	Joback Method
dvisc	0.0029863	Paxs	330.73	Joback Method
dvisc	0.0013948	Paxs	375.91	Joback Method
dvisc	0.0007671	Paxs	421.08	Joback Method
dvisc	0.0004737	Paxs	466.26	Joback Method
dvisc	0.0003185	Paxs	511.44	Joback Method
dvisc	0.0002284	Paxs	556.61	Joback Method
dvisc	0.0001722	Paxs	601.79	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C80418811&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C80418811&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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