

2-Imidazoline, 2-benzhydroxymethyl

Other names:	Diphenazoline
Inchi:	InChI=1S/C17H18N2O/c1-3-7-14(8-4-1)17(15-9-5-2-6-10-15)20-13-16-18-11-12-19-16/h
InchiKey:	RTZMBWHMKICABA-UHFFFAOYSA-N
Formula:	C17H18N2O
SMILES:	<chem>c1ccc(C(OCC2=NCCN2)c2ccccc2)cc1</chem>
Mol. weight [g/mol]:	266.34
CAS:	6703-39-5

Physical Properties

Property code	Value	Unit	Source
gf	478.72	kJ/mol	Joback Method
hf	177.26	kJ/mol	Joback Method
hfus	33.96	kJ/mol	Joback Method
hvap	74.50	kJ/mol	Joback Method
log10ws	-3.54		Crippen Method
logp	2.794		Crippen Method
mcvol	213.540	ml/mol	McGowan Method
pc	2665.27	kPa	Joback Method
rinpol	2210.00		NIST Webbook
rinpol	2275.00		NIST Webbook
rinpol	2210.00		NIST Webbook
tb	790.04	K	Joback Method
tc	1057.46	K	Joback Method
tf	546.41	K	Joback Method
vc	0.797	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	636.94	J/molxK	790.04	Joback Method
cpg	655.07	J/molxK	834.61	Joback Method
cpg	671.32	J/molxK	879.18	Joback Method
cpg	685.77	J/molxK	923.75	Joback Method
cpg	698.49	J/molxK	968.32	Joback Method

cpg	709.56	J/mol×K	1012.89	Joback Method
cpg	719.04	J/mol×K	1057.46	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C6703395&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpolar:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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