

Glutaric acid, 2-chloro-6-fluorophenyl hexyl ester

Inchi:	InChI=1S/C17H22ClFO4/c1-2-3-4-5-12-22-15(20)10-7-11-16(21)23-17-13(18)8-6-9-14(19)
InchiKey:	YDRRKXAGOQPLMQ-UHFFFAOYSA-N
Formula:	C17H22ClFO4
SMILES:	CCCCCCOC(=O)CCCC(=O)Oc1c(F)cccc1Cl
Mol. weight [g/mol]:	344.81

Physical Properties

Property code	Value	Unit	Source
gf	-489.17	kJ/mol	Joback Method
hf	-882.07	kJ/mol	Joback Method
hfus	45.90	kJ/mol	Joback Method
hvap	78.92	kJ/mol	Joback Method
log10ws	-5.43		Crippen Method
logp	4.678		Crippen Method
mcvol	255.520	ml/mol	McGowan Method
pc	1551.22	kPa	Joback Method
rinpola	2301.00		NIST Webbook
rinpola	2301.00		NIST Webbook
tb	814.28	K	Joback Method
tc	1014.35	K	Joback Method
tf	507.64	K	Joback Method
vc	0.995	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.88	J/mol×K	814.28	Joback Method
cpg	751.62	J/mol×K	847.62	Joback Method
cpg	764.38	J/mol×K	880.97	Joback Method
cpg	776.17	J/mol×K	914.31	Joback Method
cpg	787.02	J/mol×K	947.66	Joback Method
cpg	796.92	J/mol×K	981.00	Joback Method
cpg	805.90	J/mol×K	1014.35	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U391514&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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