

Succinic acid, 2,2,3,3-tetrafluoropropyl 3-nitrobenzyl ester

Inchi:	InChI=1S/C14H13F4NO6/c15-13(16)14(17,18)8-25-12(21)5-4-11(20)24-7-9-2-1-3-10(6-9
InchiKey:	FQTVEUWYOIOYMZ-UHFFFAOYSA-N
Formula:	C14H13F4NO6
SMILES:	O=C(CCC(=O)OCC(F)(F)C(F)F)OCc1cccc([N+](=O)[O-])c1
Mol. weight [g/mol]:	367.25

Physical Properties

Property code	Value	Unit	Source
gf	-1041.35	kJ/mol	Joback Method
hf	-1406.06	kJ/mol	Joback Method
hfus	43.99	kJ/mol	Joback Method
hvap	79.65	kJ/mol	Joback Method
log10ws	-4.28		Crippen Method
logp	2.862		Crippen Method
mcvol	223.740	ml/mol	McGowan Method
pc	1898.60	kPa	Joback Method
rinpol	2219.00		NIST Webbook
rinpol	2219.00		NIST Webbook
tb	849.21	K	Joback Method
tc	1059.85	K	Joback Method
tf	564.19	K	Joback Method
vc	0.896	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	674.23	J/molxK	849.21	Joback Method
cpg	684.45	J/molxK	884.32	Joback Method
cpg	693.72	J/molxK	919.42	Joback Method
cpg	702.07	J/molxK	954.53	Joback Method
cpg	709.55	J/molxK	989.64	Joback Method
cpg	716.19	J/molxK	1024.74	Joback Method
cpg	722.03	J/molxK	1059.85	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U390176&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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