

# Fumaric acid, monoamide, N-(2-bromophenyl)-, 2-fluorophenyl ester

**Inchi:** InChI=1S/C16H11BrFNO3/c17-11-5-1-3-7-13(11)19-15(20)9-10-16(21)22-14-8-4-2-6-12

**InchiKey:** FAKJERTWWPSRBD-MDZDMXLPSA-N

**Formula:** C16H11BrFNO3

**SMILES:** O=C(C=CC(=O)Oc1ccccc1F)Nc1ccccc1Br

**Mol. weight [g/mol]:** 364.17

## Physical Properties

Property code	Value	Unit	Source
gf	-84.32	kJ/mol	Joback Method
hf	-279.92	kJ/mol	Joback Method
hfus	42.55	kJ/mol	Joback Method
hvap	85.00	kJ/mol	Joback Method
log10ws	-4.99		Crippen Method
logp	3.688		Crippen Method
mvol	222.740	ml/mol	McGowan Method
pc	2701.41	kPa	Joback Method
rinpol	2759.00		NIST Webbook
rinpol	2759.00		NIST Webbook
tb	878.72	K	Joback Method
tc	1124.02	K	Joback Method
tf	578.02	K	Joback Method
vc	0.841	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	593.11	J/mol×K	878.72	Joback Method
cpg	603.29	J/mol×K	919.60	Joback Method
cpg	612.52	J/mol×K	960.49	Joback Method
cpg	620.91	J/mol×K	1001.37	Joback Method
cpg	628.52	J/mol×K	1042.26	Joback Method
cpg	635.45	J/mol×K	1083.14	Joback Method
cpg	641.77	J/mol×K	1124.02	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357480&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357480&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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