

# Succinic acid, 1,1,1-trifluoroprop-2-yl 4-methoxybenzyl ester

<b>Inchi:</b>	InChI=1S/C15H17F3O5/c1-10(15(16,17)18)23-14(20)8-7-13(19)22-9-11-3-5-12(21-2)6-4
<b>InchiKey:</b>	HNCHDMOJDHTNGL-UHFFFAOYSA-N
<b>Formula:</b>	C15H17F3O5
<b>SMILES:</b>	COc1ccc(COC(=O)CCC(=O)OC(C)C(F)(F)F)cc1
<b>Mol. weight [g/mol]:</b>	334.29

## Physical Properties

Property code	Value	Unit	Source
gf	-978.67	kJ/mol	Joback Method
hf	-1352.05	kJ/mol	Joback Method
hfus	33.32	kJ/mol	Joback Method
hvap	68.51	kJ/mol	Joback Method
log10ws	-3.90		Crippen Method
logp	3.013		Crippen Method
mvol	224.510	ml/mol	McGowan Method
pc	1772.85	kPa	Joback Method
rinpol	1912.00		NIST Webbook
rinpol	1912.00		NIST Webbook
tb	743.40	K	Joback Method
tc	937.13	K	Joback Method
tf	453.49	K	Joback Method
vc	0.871	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	646.98	J/mol×K	743.40	Joback Method
cpg	660.30	J/mol×K	775.69	Joback Method
cpg	672.71	J/mol×K	807.98	Joback Method
cpg	684.21	J/mol×K	840.26	Joback Method
cpg	694.83	J/mol×K	872.55	Joback Method
cpg	704.57	J/mol×K	904.84	Joback Method
cpg	713.47	J/mol×K	937.13	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389688&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389688&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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