

# 1,3,5-tris-(Bromomethyl) cyclohexane

<b>Inchi:</b>	InChI=1S/C9H15Br3/c10-4-7-1-8(5-11)3-9(2-7)6-12/h7-9H,1-6H2
<b>InchiKey:</b>	HPDNDWLWKJEJTR-UHFFFAOYSA-N
<b>Formula:</b>	C9H15Br3
<b>SMILES:</b>	BrCC1CC(CBr)CC(CBr)C1
<b>Mol. weight [g/mol]:</b>	362.93

## Physical Properties

Property code	Value	Unit	Source
gf	76.89	kJ/mol	Joback Method
hf	-136.46	kJ/mol	Joback Method
hfus	28.90	kJ/mol	Joback Method
hvap	54.74	kJ/mol	Joback Method
log10ws	-4.05		Crippen Method
logp	4.204		Crippen Method
mcvol	179.310	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
rinpol	1937.00		NIST Webbook
rinpol	1937.00		NIST Webbook
tb	614.01	K	Joback Method
tc	858.93	K	Joback Method
tf	369.49	K	Joback Method
vc	0.656	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	395.57	J/molxK	614.01	Joback Method
cpg	411.92	J/molxK	654.83	Joback Method
cpg	427.09	J/molxK	695.65	Joback Method
cpg	441.14	J/molxK	736.47	Joback Method
cpg	454.14	J/molxK	777.29	Joback Method
cpg	466.15	J/molxK	818.11	Joback Method
cpg	477.24	J/molxK	858.93	Joback Method
dvisc	0.0019928	Paxs	369.49	Joback Method

dvisc	0.0012818	Paxs	410.24	Joback Method
dvisc	0.0008930	Paxs	451.00	Joback Method
dvisc	0.0006605	Paxs	491.75	Joback Method
dvisc	0.0005116	Paxs	532.50	Joback Method
dvisc	0.0004110	Paxs	573.26	Joback Method
dvisc	0.0003398	Paxs	614.01	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R96333&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R96333&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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