

# Chloromethanesulfonyl chloride

<b>Other names:</b>	Methanesulfonyl chloride, chloro-Chlorid kyseliny chlormethansulfonove Chlormethansulfochlorid Chlormethansulfonylchlorid chloromethanesulphonyl chloride
<b>Inchi:</b>	InChI=1S/CH2Cl2O2S/c2-1-6(3,4)5/h1H2
<b>InchiKey:</b>	KQDDQXNVESLJNO-UHFFFAOYSA-N
<b>Formula:</b>	CH2Cl2O2S
<b>SMILES:</b>	O=S(=O)(Cl)CCl
<b>Mol. weight [g/mol]:</b>	149.00
<b>CAS:</b>	3518-65-8

## Physical Properties

Property code	Value	Unit	Source
gf	-534.86	kJ/mol	Joback Method
hf	-548.80	kJ/mol	Joback Method
hfus	18.12	kJ/mol	Joback Method
hvap	45.22	kJ/mol	Joback Method
log10ws	-0.87		Crippen Method
logp	0.751		Crippen Method
mcpol	77.520	ml/mol	McGowan Method
pc	6577.70	kPa	Joback Method
rinpol	904.00		NIST Webbook
rinpol	904.00		NIST Webbook
rinpol	910.00		NIST Webbook
tb	344.92	K	Joback Method
tc	526.50	K	Joback Method
tf	199.43	K	Joback Method
vc	0.316	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	108.84	J/molxK	344.92	Joback Method

cpg	112.61	J/mol×K	375.18	Joback Method
cpg	116.27	J/mol×K	405.45	Joback Method
cpg	119.83	J/mol×K	435.71	Joback Method
cpg	123.27	J/mol×K	465.97	Joback Method
cpg	126.58	J/mol×K	496.23	Joback Method
cpg	129.76	J/mol×K	526.50	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3518658&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3518658&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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