

# 2-Hexene, 5,5-dimethyl-, (Z)-

<b>Other names:</b>	(2Z)-5,5-Dimethyl-2-hexene (Z)-5,5-Dimethylhex-2-ene 5,5-Dimethyl-2-hexene (cis) 5,5-Dimethyl-2-hexene, (2Z)- 5,5-Dimethyl-cis-2-hexene cis-2-Hexene, 5,5-dimethyl cis-5,5-Dimethyl-2-Hexene
<b>Inchi:</b>	InChI=1S/C8H16/c1-5-6-7-8(2,3)4/h5-6H,7H2,1-4H3/b6-5-
<b>InchiKey:</b>	NWZJLSKAFZXSQH-WAYWQWQTSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	CC=CCC(C)(C)C
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	39761-61-0

## Physical Properties

Property code	Value	Unit	Source
gf	99.54	kJ/mol	Joback Method
hf	-99.98	kJ/mol	Joback Method
hfus	9.26	kJ/mol	Joback Method
hvap	38.10	kJ/mol	NIST Webbook
log10ws	-2.78		Crippen Method
logp	2.999		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2726.86	kPa	Joback Method
rinpol	721.80		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	724.00		NIST Webbook
rinpol	721.50		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	709.00		NIST Webbook
rinpol	732.50		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	723.00		NIST Webbook
rinpol	732.50		NIST Webbook
rinpol	723.50		NIST Webbook
ripol	757.00		NIST Webbook

tb	380.10 ± 0.50	K	NIST Webbook
tc	566.77	K	Joback Method
tf	177.26	K	Joback Method
vc	0.453	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	219.13	J/mol×K	383.37	Joback Method
cpg	284.31	J/mol×K	536.21	Joback Method
cpg	272.68	J/mol×K	505.64	Joback Method
cpg	260.39	J/mol×K	475.07	Joback Method
cpg	247.38	J/mol×K	444.50	Joback Method
cpg	233.64	J/mol×K	413.94	Joback Method
cpg	295.30	J/mol×K	566.77	Joback Method
dvisc	0.0002263	Paxs	383.37	Joback Method
dvisc	0.0003144	Paxs	349.02	Joback Method
dvisc	0.0004693	Paxs	314.67	Joback Method
dvisc	0.0007728	Paxs	280.31	Joback Method
dvisc	0.0014626	Paxs	245.96	Joback Method
dvisc	0.0034054	Paxs	211.61	Joback Method
dvisc	0.0110015	Paxs	177.26	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.25753e+01
Coeff. B	-2.60996e+03
Coeff. C	-6.53430e+01
Temperature range (K), min.	277.75
Temperature range (K), max.	424.65

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C39761610&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C39761610&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>d<sub>visc</sub>:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pv<sub>ap</sub>:</b>	Vapor pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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