

# 2-Methyl-1,2-dibromobutane

Inchi:	InChI=1S/C5H10Br2/c1-3-5(2,7)4-6/h3-4H2,1-2H3
InchiKey:	GWFLFLQACQDTLD-UHFFFAOYSA-N
Formula:	C5H10Br2
SMILES:	CCC(C)(Br)CBr
Mol. weight [g/mol]:	229.94

## Physical Properties

Property code	Value	Unit	Source
gf	22.70	kJ/mol	Joback Method
hf	-102.62	kJ/mol	Joback Method
hfus	11.86	kJ/mol	Joback Method
hvap	38.30	kJ/mol	Joback Method
log10ws	-2.89		Crippen Method
logp	2.945		Crippen Method
mcvol	116.310	ml/mol	McGowan Method
pc	4211.09	kPa	Joback Method
rinpol	994.00		NIST Webbook
tb	442.89	K	Joback Method
tc	661.07	K	Joback Method
tf	268.13	K	Joback Method
vc	0.428	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	197.85	J/molxK	442.89	Joback Method
cpg	208.11	J/molxK	479.25	Joback Method
cpg	217.61	J/molxK	515.62	Joback Method
cpg	226.38	J/molxK	551.98	Joback Method
cpg	234.49	J/molxK	588.34	Joback Method
cpg	242.00	J/molxK	624.70	Joback Method
cpg	248.94	J/molxK	661.07	Joback Method
dvisc	0.0043647	Paxs	268.13	Joback Method
dvisc	0.0024400	Paxs	297.26	Joback Method

dvisc	0.0015133	Paxs	326.38	Joback Method
dvisc	0.0010149	Paxs	355.51	Joback Method
dvisc	0.0007231	Paxs	384.64	Joback Method
dvisc	0.0005404	Paxs	413.76	Joback Method
dvisc	0.0004197	Paxs	442.89	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R559390&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R559390&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/12-099-9/2-Methyl-1-2-dibromobutane.pdf>

Generated by Cheméo on 2024-04-19 22:31:33.450184097 +0000 UTC m=+15855142.370761408.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.